

Chapter 24: Lipids. Hydrophobic (non-polar, soluble in organic solvent), typically of low molecular weight compounds of organic origin.

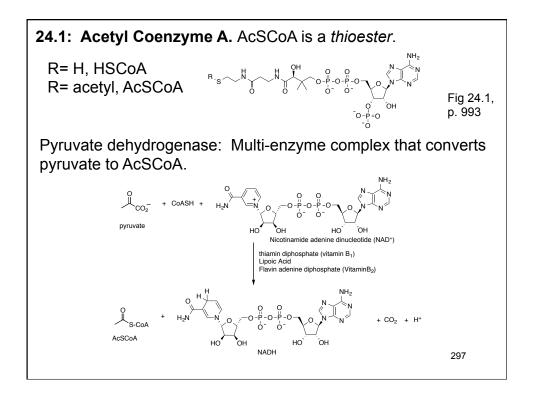
- fatty acids and waxes
- essential oils
- many vitamins
- hormones (non-peptide)
- components of cell membranes (non-peptide)

Share a common biosynthesis that ultimately derives their carbon source from glucose (glycolysis)

Glucose \rightarrow pyruvate \rightarrow lactate

 $\begin{array}{c} CHO \\ H \longrightarrow OH \\ HO \longrightarrow H \\ H \longrightarrow OH \\ H \longrightarrow OH \\ CH_2OH \end{array} \xrightarrow{\begin{array}{c} Q \\ Q \\ Q \\ H \end{array} \xrightarrow{\begin{array}{c} Q \\ Q \\ Q \\ H \end{array}} \xrightarrow{\begin{array}{c} Q \\ Q \\ Q \\ Q \\ H \end{array} \xrightarrow{\begin{array}{c} Q \\ Q \\ Q \\ H \end{array}} \xrightarrow{\begin{array}{c} Q \\ Q \\ Q \\ Q \\ H \end{array} \xrightarrow{\begin{array}{c} Q \\ Q \\ Q \\ H \end{array}} \xrightarrow{\begin{array}{c} Q \\ Q \\ Q \\ Q \\ H \end{array} \xrightarrow{\begin{array}{c} Q \\ Q \\ Q \\ H \end{array} \xrightarrow{\begin{array}{c} Q \\ Q \\ Q \\ H \end{array}} \xrightarrow{\begin{array}{c} Q \\ Q \\ Q \\ Q \\ H \end{array} \xrightarrow{\begin{array}{c} Q \\ Q \\ Q \\ H \end{array} \xrightarrow{\begin{array}{c} Q \\ Q \\ Q \\ H \end{array}} \xrightarrow{\begin{array}{c} Q \\ Q \\ Q \\ H \end{array} \xrightarrow{\begin{array}{c} Q \\ Q \\ Q \\ H \end{array} \xrightarrow{\begin{array}{c} Q \\ Q \\ Q \\ H \end{array} \xrightarrow{\begin{array}{c} Q \\ Q \\ H \end{array} \xrightarrow{\begin{array}{c} Q \\ Q \\ Q \\ H \end{array}} \xrightarrow{\begin{array}{c} Q \\ Q \\ Q \\ H \end{array} \xrightarrow{\begin{array}{c} Q \\ Q \\ Q \\ H \end{array} \xrightarrow{\begin{array}{c} Q \\ H \end{array} \xrightarrow{\begin{array}{c} Q \\ Q \end{array} \xrightarrow{\begin{array}{c} Q \\ Q \\ H \end{array} \xrightarrow{\begin{array}{c} Q \\ Q \\ H \end{array} \xrightarrow{\begin{array}{c} Q \\ Q \end{array} \xrightarrow{\begin{array}{c} Q \\ Q \\ H \end{array}\xrightarrow{\begin{array}{c} Q \\ Q \end{array} \xrightarrow{\begin{array}{c} Q \\ Q \end{array}}\xrightarrow{\begin{array}{c} Q \\ Q \end{array} \xrightarrow{\begin{array}{c} Q \\ Q \end{array}}\xrightarrow{\begin{array}{c} Q \\ Q \end{array} \xrightarrow{\begin{array}{c} Q \\ \end{array}}\xrightarrow{\begin{array}{c} Q \\ \end{array}}\xrightarrow{\begin{array}{c} Q \\ \end{array}\xrightarrow{\begin{array}{c} Q \\ Q \end{array}\xrightarrow{\begin{array}{c} Q \end{array} \xrightarrow{\begin{array}{c} Q \\ Q \end{array}}\xrightarrow{\begin{array}{c} Q \\ \end{array}}\xrightarrow{\begin{array}{c} Q \end{array} \xrightarrow{\begin{array}{c} Q \\ \end{array}}\xrightarrow{\begin{array}{c} Q \\ \end{array}}\xrightarrow{\begin{array}{c} Q \end{array} \xrightarrow{\begin{array}{c} Q \\ \end{array}}\xrightarrow{\begin{array}{c} Q \end{array}\end{array}\xrightarrow{\begin{array}{c} Q \end{array} \xrightarrow{\begin{array}{c} Q \end{array}}\xrightarrow{\begin{array}{c} Q \end{array}\end{array}\xrightarrow{\begin{array}{c} Q \end{array}}\xrightarrow{\begin{array}{c} Q \end{array}\end{array}\xrightarrow{\begin{array}{c} Q \end{array}\end{array}\xrightarrow{\begin{array}{c} Q \end{array}\end{array}\xrightarrow{\begin{array}{c} Q \end{array}}\xrightarrow{\begin{array}{c} Q \end{array}\end{array}\xrightarrow{\begin{array}{c} Q \end{array}\end{array}\xrightarrow{\begin{array}{c} Q \end{array}}\xrightarrow{\begin{array}{c} Q \end{array}}\xrightarrow{\begin{array}{c} Q \end{array}\end{array}\xrightarrow{\begin{array}{c} Q$

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Acetyl CoA is a thioester. Thioesters are more reactive toward nucleophilic acyl substitution than esters, but considerably less reactive than acid chlorides and anhydrides. + CoASH + Nu-H choline acetyltransferase + CoASH N(CH₃)₃ N(CH₃)₃ HO acetylcholine choline Thioester enolize more readily than esters. The enol can react with electrophile to afford α -substitution products E S-CoA $\mathcal{A}_{S-CoA} \longrightarrow \mathcal{A}_{S-CoA}$ O S-CoA acetyl-CoA carboxylase + HCO₃⁻ biotin (vitamin B₇), - 0 S-CoA + H₂O Malonyl CoA 298

