Chemistry 220b-01 Exam 2 (100 pts) Tuesday, March 1, 2011 Chapters 12 – 16

Name ____

Write and sign the VU Honor Pledge:

This exam is closed book and closed notes

signature

NOTE: It is difficult for me to give you partial credit if you do not show your work!

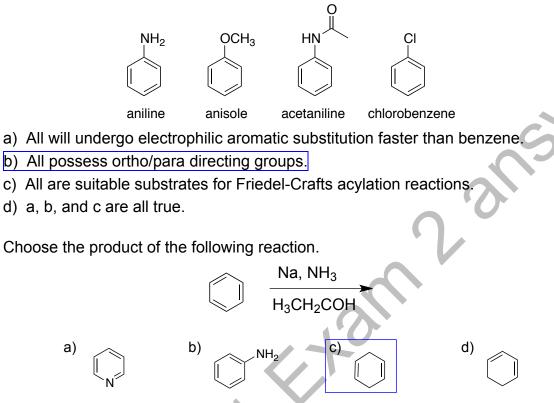
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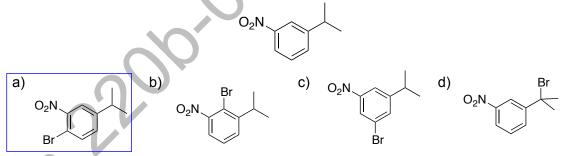
Good Luck !!

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- 1-14. Multiple choice. Choose the *best* answer for the following questions (4 pts each, 56 pts)
- 1. Which of the following statements about aniline, anisole, acetaniline, and chlorobenzene is true?

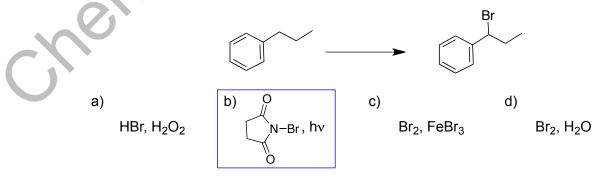


3. Predict the *major* product from the electrophilic bromination of 3-nitro-(1-methylethyl)benzene (shown below).



4. Choose the best reagent for the following reaction.

2.

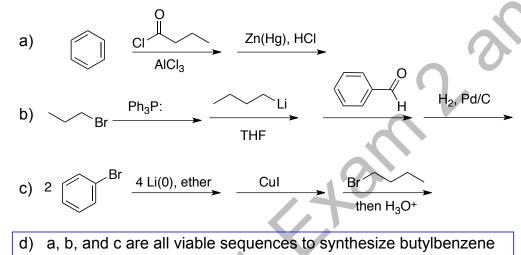


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5. Which reagent or sequence of reagents will convert benzenediazonium chloride to cyanobenzene?



- a) KCN, AICI3
- b) NH₃
- c) CuCN
- d) 1) NBS; 2) NaCN, DMSO
- 6. Which of the following sequences is a viable way to synthesize butylbenzene.



7 Which is the best reagent for the following reaction?

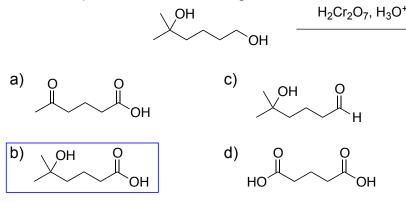
a) H₂, Pd/C

- b) LiAlH₄, ether; then HCl
- c) Zn, H_3O^+
- d) a, b, and c will work equally well
- 8. Which reagent or sequence of reagents can be used to synthesize aniline from benzene?
 - a) H_2NNH_2 , KOH, Δ
 - b) 1) Br₂, FeBr₃; 2) NH₃
 - c) 1) NaNO₂, HCl; 2) Sn, HCl
 - d) 1) HNO₃, H₂SO₄; 2) H₂, Pd/C

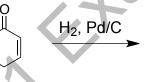
Name

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9. Predict the product of the following reaction.



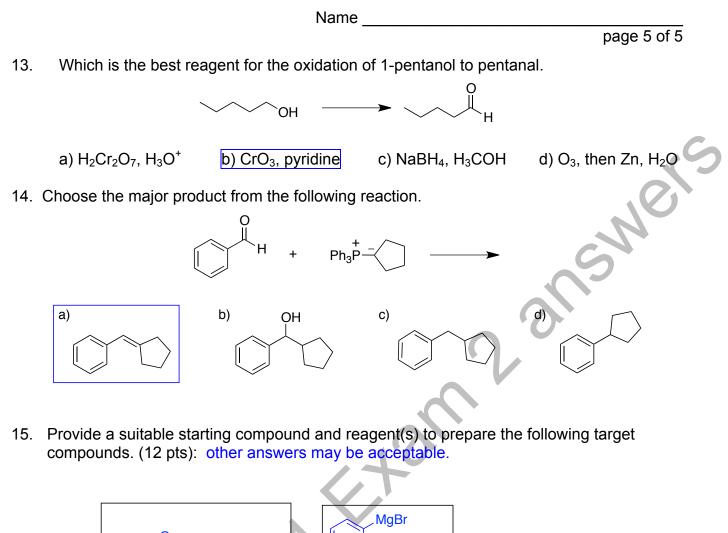
- 10. The most diagnostic spectroscopic method for the presence of an aldehyde is . . .
 - a) a ¹H NMR resonance between δ = 9 10 ppm.
 - b) a ¹³C NMR resonance between δ = 165 220 ppm.
 - c) an infrared absorbance between $1700 1750 \text{ cm}^{-1}$.
 - d) a weak UV absorbance at 260 nm with a small extinction coefficient (ϵ).
- 11. The carbonyl absorbance in the IR spectrum of cyclohexenone is at 1685 cm⁻¹. Upon catalytic hydrogenation of cyclohexenone, the carbonyl absorbance of the product . . .

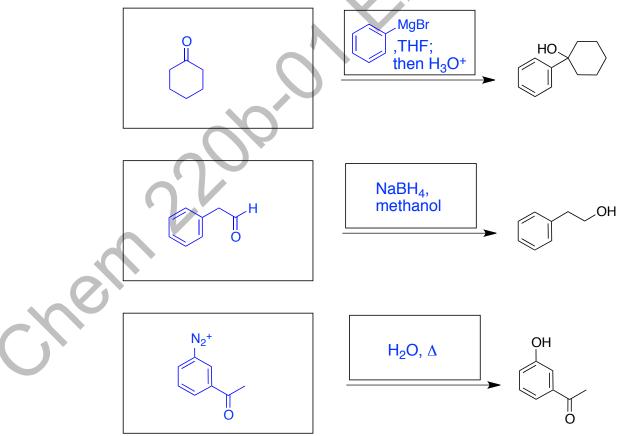


- a) . . . is expected at 1685 cm⁻¹
- b) ... is expected at 1655 cm⁻¹
- c) . . . is expected at 1715 cm⁻¹.
- d) . . . is expected to disappear and a new broad absorbance between 3000-3500 cm⁻¹ will be present.
- 12. The equilibrium constant for the hydration of butanal is 0.6. The addition of an acid catalyst . . .

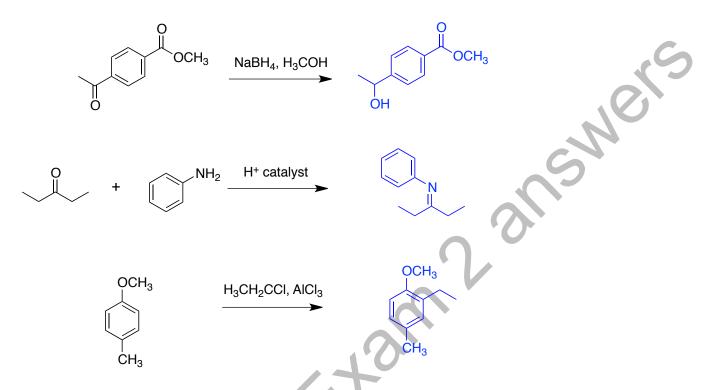


- a) \ldots will dramatically shift the equilibrium toward the hydrate (K_{eq} is very large).
- b) . . . will dramatically shift the equilibrium toward butanal (K_{eq} is very small).
- c) . . . will result in a 50:50 mixture of butanal and its hydrate ($K_{eq} = 1$).
- d) . . . does not change the equilibrium ($K_{eq} = 0.6$).

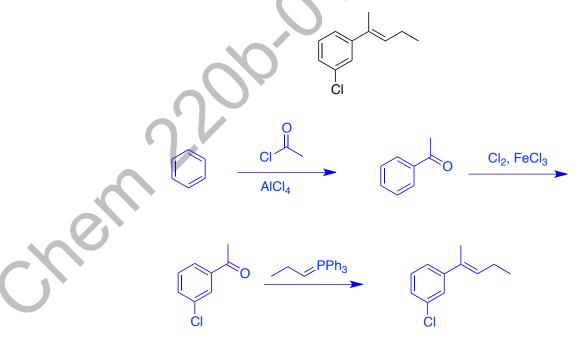




16. Provide the *major* product of the following reactions. (12 pts)



17. Synthesize the target compound below starting from benzene, any <u>primary alkyl</u> halide (including iodomethane), any organometallic reagent or Wittig reagent derived from a <u>primary alkyl</u> halide (including iodomethane), any acid chloride, and any inorganic reagents. *Remember, brevity is the soul of wit and organic synthesis* (10 pts)



other answers may be acceptable

18. Provide a complete mechanism for the acid-catalyzed reaction of cyclohexanone and methanol to afford 1,1-dimethoxycyclohexane. (10 pts)

