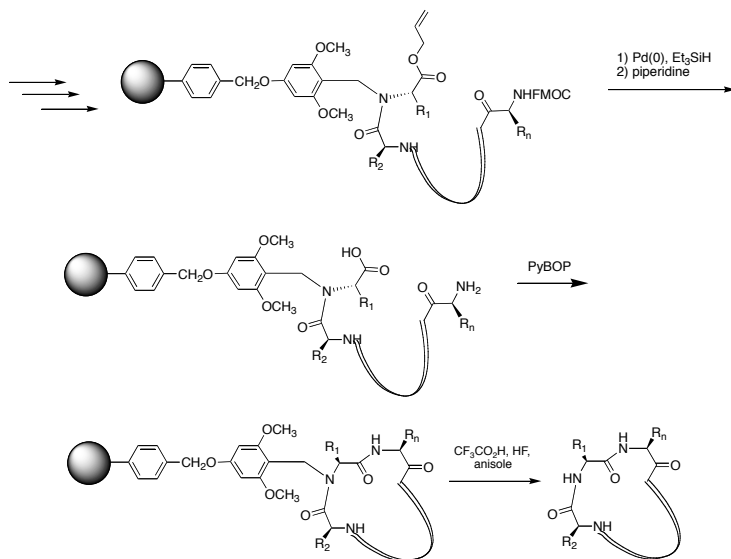


On-support cyclization

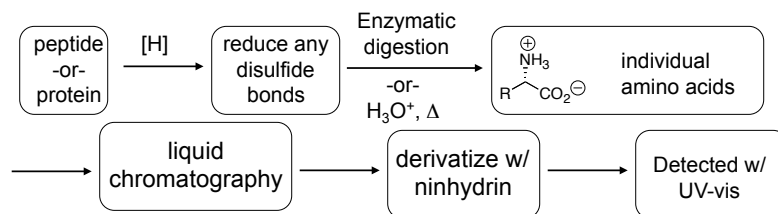


49

Peptide and Protein Analysis

Primary (1°) structure of a peptide or protein is the amino acid sequence

Amino acid analyzer- automated instrument to determine the amino acid content of a peptide or protein. Individual amino acids are separated by hplc, then detected by post-column derivatization

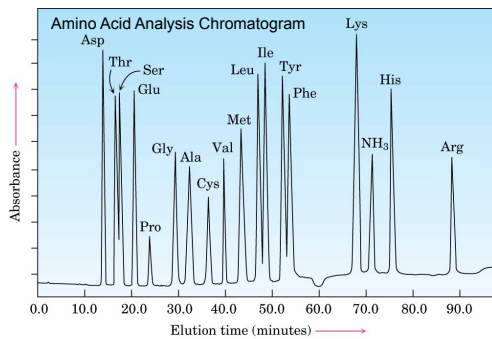
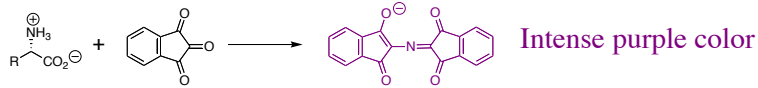


Different amino acids have different chromatographic mobilities (retention times)

1972 Nobel Prize in Chemistry
William Stein
Stanford Moore

50

Reaction of primary amines with ninhydrin



So, why is it necessary to use a post- rather than pre-column derivatization protocol?

Why are there only 17 AA's in the chromatogram?

51

Fluorescence Detection- less background, greater sensitivity, lower detection limits

Absorption spectroscopy- wavelength that light absorbs, molecules are in an electronically excited state

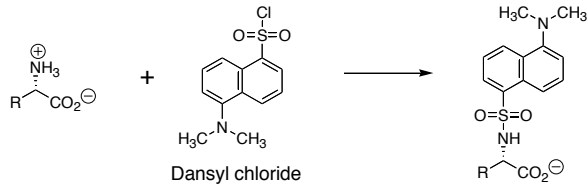
Emission spectroscopy- the excited molecules relax by emission of a photon.

Fluorescence- excitation wavelength and emission wavelength are different. Molecule will emit light at longer (lower energy) wavelength than is absorbs.

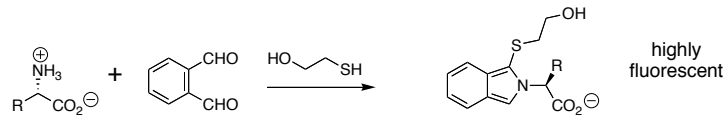
52

Fluorescent tags

Dansyl- detected by UV or fluorescence

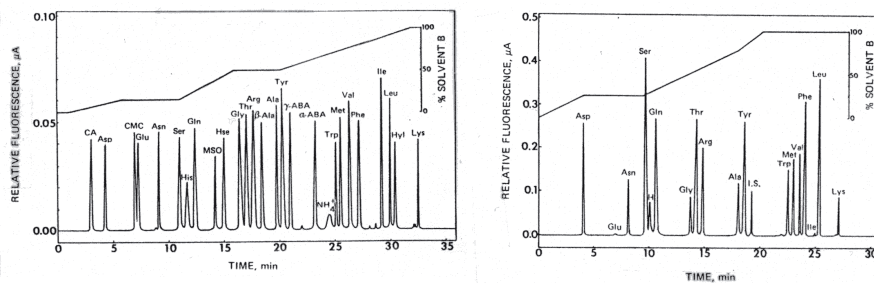


OPA (o-phthalaldehyde)- detected by fluorescence



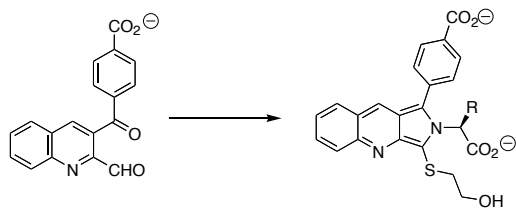
53

Reversed-phase (C-18) HPLC Trace 5 pmols amino acids w/ OPA, HOCH₂CH₂SH



54

Attomol detection w/ laser induced fluorescence



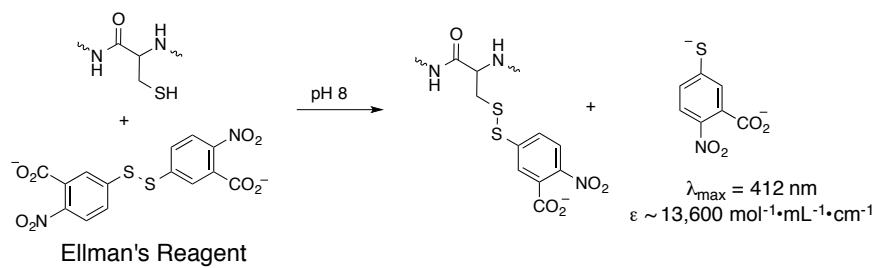
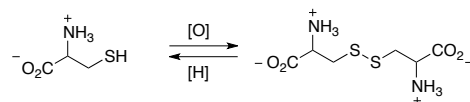
excitation: 488 nm
emission: 560 nm

10^{-3} milli
 10^{-6} micro
 10^{-9} nano
 10^{-12} pico
 10^{-15} fempto
 10^{-18} atto
 10^{-21} zepto

Avagadro' s number 10^{23}

55

Cysteine vs. Cystine



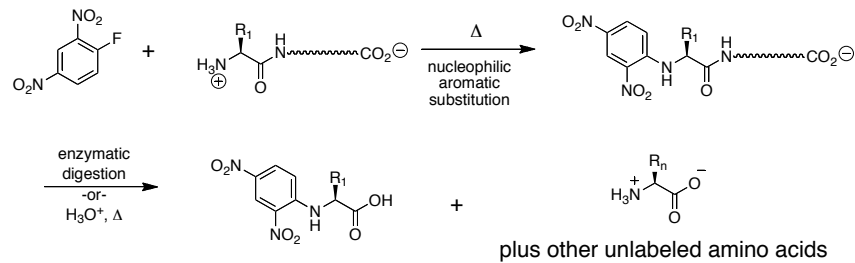
56

Peptide and Protein Sequences:

primary (1°) structure- amino acid sequence

N-labeling with Sanger's reagent: Sanger's (2,4-dinitrofluorobenzene)

reagent reacts with the N-terminal amino group and has a diagnostic UV absorbance that is detected after enzymatic digestion and amino acid analysis

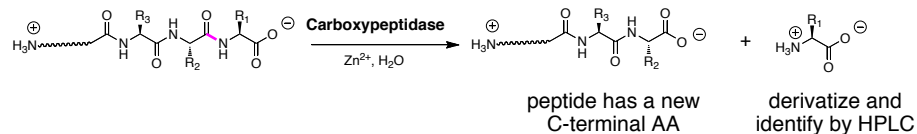


N-terminal amino acid is specifically labeled with a unique UV chromophore

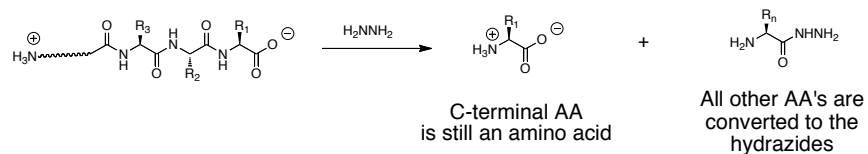
57

C-terminal sequencing:

Carboxypeptidase- enzyme that hydrolyzed amide bonds of a peptide or protein starting from the C-terminal end (exopeptidase)



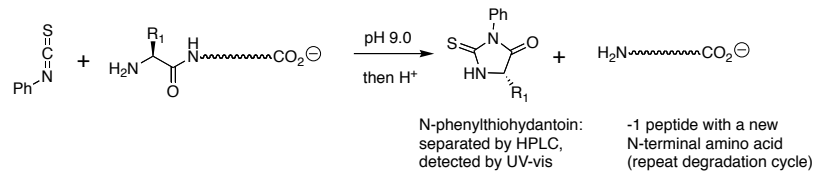
Hydrolyze peptide with hydrazine ($\text{H}_2\text{N}-\text{NH}_2$)



58

Edman Degradation: chemical method for the sequential cleavage and identification of the amino acids of a peptide, one at a time starting from the N-terminus.

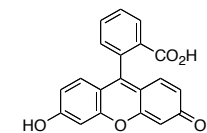
Reagent: Ph-N=C=S, phenylisothiocyanate



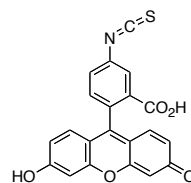
59

Peptide sequencing by Edman degradation: Monitor the appearance of N-phenylthiohydantoin over time to get the peptide sequence. Good for peptides up to ~ 25 amino acids long. Longer peptides and proteins must be cut into smaller fragments before Edman sequencing

Fluorescent Edman sequencing reagent



Fluorescein
(a common fluorescent dye)



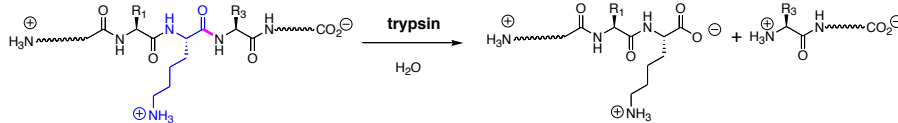
Fluorescein Isothiocyanate
(a fluorescent Edman reagent)

60

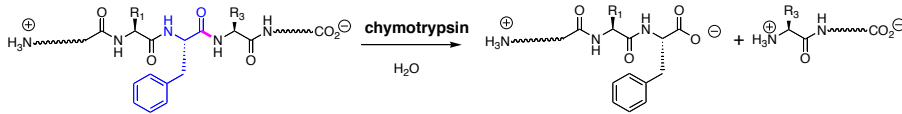
Enzymatic and chemical cleavage of peptides and proteins at defined sites

Enzymatic

- trypsin: cleaves at the C-terminal side of basic residues, Arg, Lys but not His

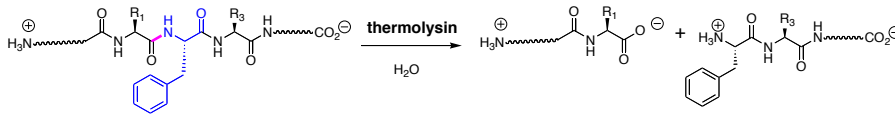


- chymotrypsin: cleaves at the C-terminal side of aromatic residues Phe, Tyr, Trp



61

- thermolysin: cleaves at the N-terminal side of hydrophobic residues Phe, Trp, Leu



Chymotrypsin cleavage products

Tyr
Asp-Asn-Gln
Gly-Gly-Phe
Leu-Arg-Arg-Ile-Arg-Pro-Lys-Leu-Lys-Trp

Trypsin cleavage products

Arg
Leu-Lys
Ile-Arg-Pro-Lys
Tyr-Gly-Gly-Phe-Leu-Arg
Trp-Asp-Asn-Gln

Trypsin: Tyr-Gly-Gly-Phe-Leu-Arg Arg Ile-Arg-Pro-Lys Leu-Lys Trp-Asp-Asn-Gln

Chymotrypsin: Tyr Gly-Gly-Phe Leu-Arg-Arg-Ile-Arg-Pro-Lys-Leu-Lys-Trp Asp-Asn-Gln

62

Other Commonly Used Protein Digest Reagents

Glu-C – cleaves to the C-terminal side of Glu residues
(cleavage at Asp is 100-300 times slower)

Asp-N – cleaves to the N-terminal side of Asp residues
and cysteic acid

Lys-N – cleaves to the N-terminal side of Lys residues

Lys-C – cleaves to the C-terminal side of Lys residues

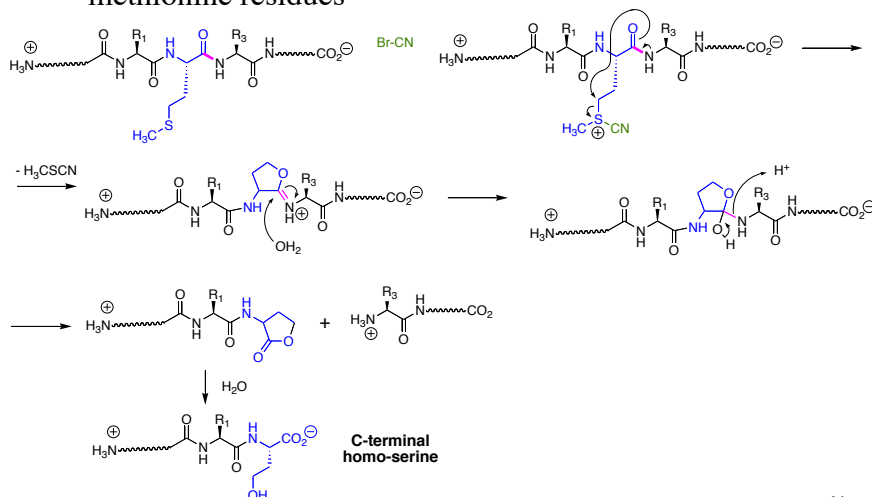
Arg-C – cleaves to the C-terminal side of Arg residues

Non-specific proteases: pepsin, proteinase K, subtilisin

63

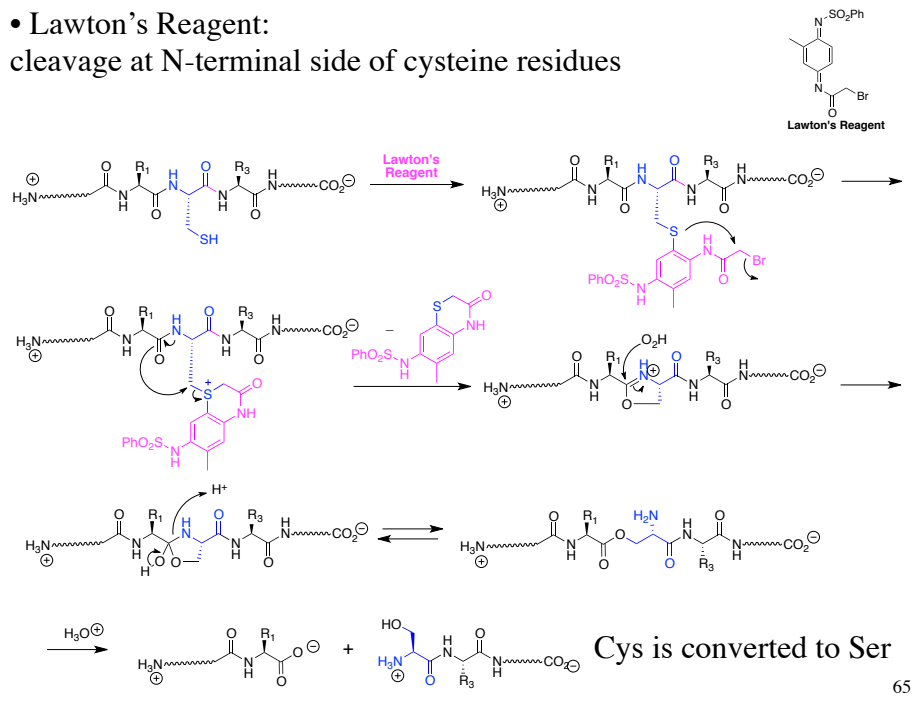
Chemical cleavage of peptides and proteins at defined sites

- Cyanogen bromide (Br-CN): cleaves to the C-terminal side of methionine residues



64

• Lawton's Reagent:
cleavage at N-terminal side of cysteine residues



EPIDERMAL GROWTH FACTOR (EGF)

H₂N-ASN1•SER2•**TYR3**•PRO4•GLY5•CYS6•PRO7•SER8•SER9•**TYR10**•
 ASP11•GLY12•**TYR13**•CYS14•LEU15•ASN16•GLY17•GLY18•VAL19•
 CYS20•**MET21**•HIS22•ILE23•GLU24•SER25•LEU26•ASP27•SER28•
TYR29•THR30•CYS31•ASN32•CYS33•VAL34•ILE35•GLY36•**TYR37**•
 SER38•GLY39•ASP40•**ARG41**•CYS42•GLN43•THR44•**ARG45**•ASP46•
 LEU47•**ARG48**•TRP49•TRP50•GLU51•LEU52•ARG53-CO₂H

Trypsin

Chymotrypsin

Cyanogen Bromide

Disulfides bridges at: **Cys6 - Cys20**
 Cys14 - Cys31
 Cys33 - Cys42

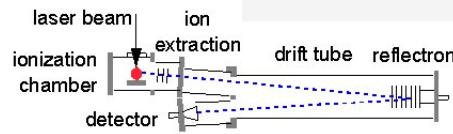
S. Cohen et al. *J. Biol. Chem.* 1972, 247, 5928-5934
 1972, 247, 7612-7621
 1973, 248, 7669-7672

66

Peptide sequencing by tandem mass spectrometry

Ionization: SIMS (secondary ion mass spectrometry)

Time-of-flight (TOF) mass spectrometer



Methods to get large, polar molecules into the gas phase for MS analysis

FAB: Fast Atom Bombardment

MALDI: Matrix-Assisted Laser Desorption Ionization

ESI: Electrospray Ionization

Mass spectrometry gives mass/charge (m/z) ratio

“Introduction to Proteomics: Tools for the New Biology,” Liebler, D. C., Humana Press: **2002**

67

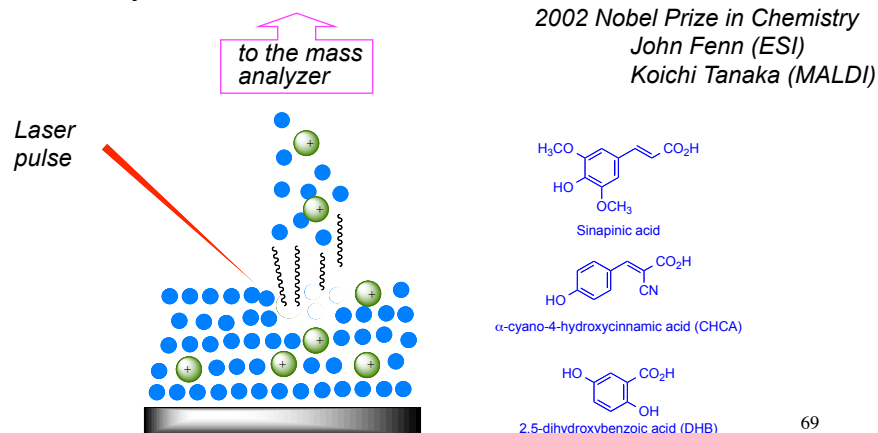
Mass spectrometry is a gas phase technique. Peptides (and proteins) are charged, polar, high molecular weight molecules (ions). How can peptides and proteins be coaxed into the gas phase?

Electrospray ionization (ESI): analyte is introduced into the mass spectrometer as an aerosol.



68

MALDI ionization (matrix-assisted laser desorption): analyte is co-crystallized with an organic molecule that has an intense UV absorption. A laser that is tuned to the absorption of the matrix, is “pulsed” at the MALDI matrix and energy is indirectly transferred to the analyte.



69

Mass Spectrometry (MS): measures the mass to charge ratio (m/z)

Dalton (Da) or mass unit (u) = units for measuring molecular masses.

One Da. = 1/12 the mass of the ^{12}C atom

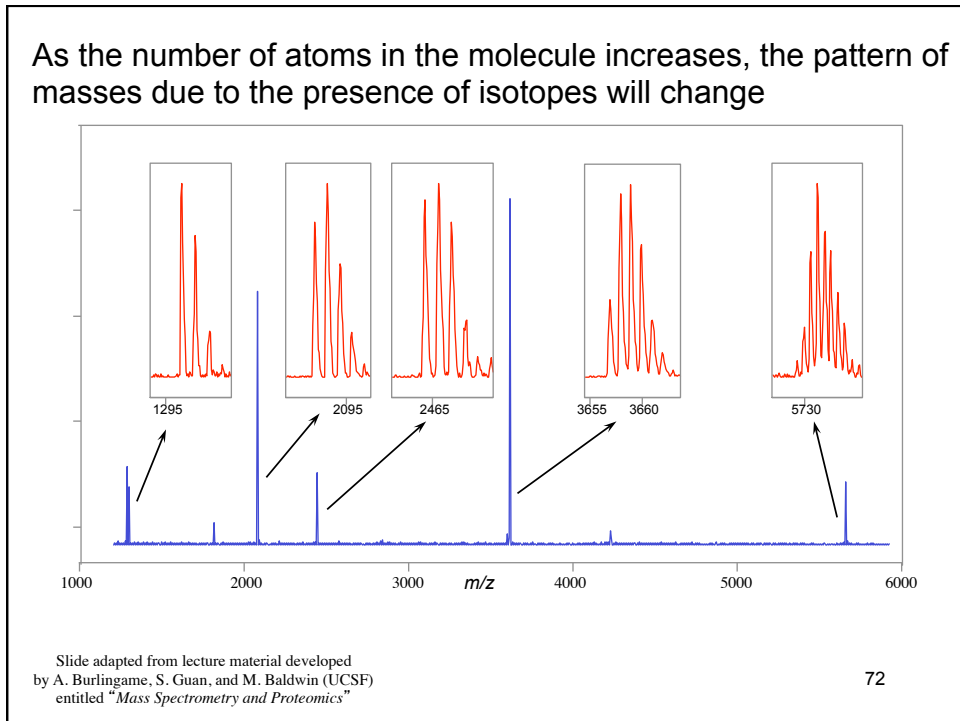
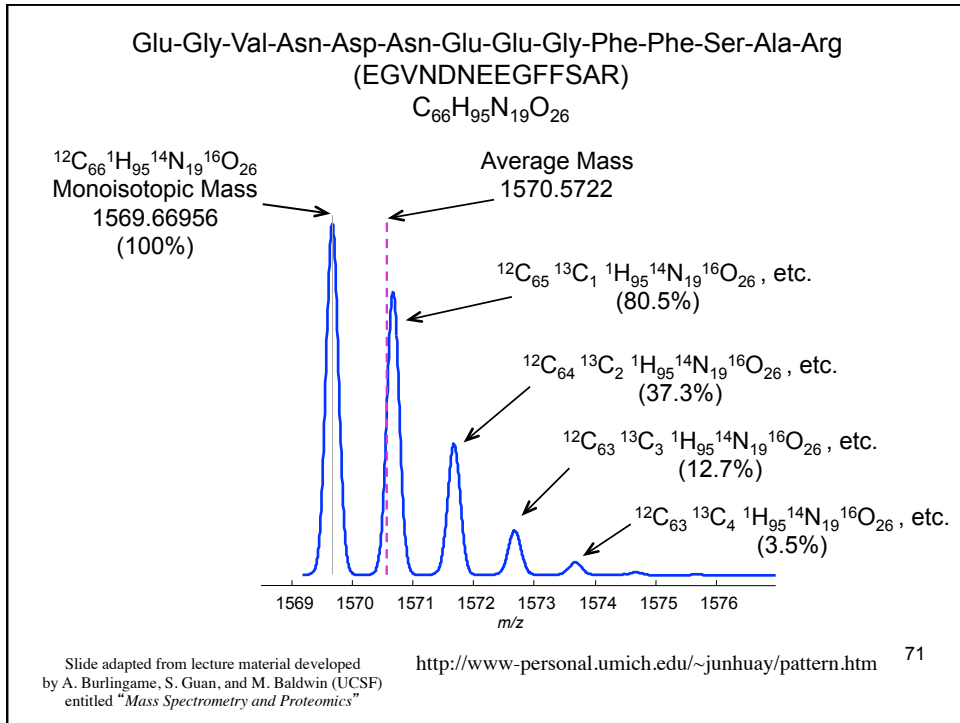
Monoisotopic mass – sum of the exact masses of the most abundant isotope of each element in a molecule

Average mass – sum of the averaged masses of each element in a molecules, weighted according to isotopic abundance.

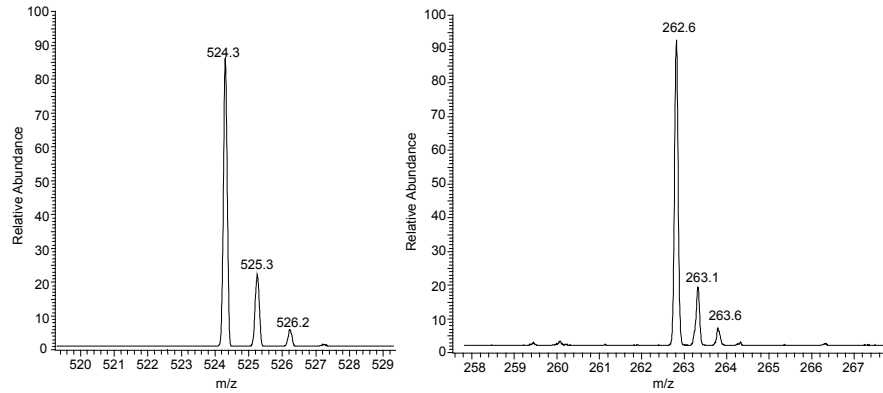
Nominal mass – mass calculated using the integer mass of the most abundant isotope for each element (H=1, C=12, O=16, N=14, etc.)

Isotope	Mass	Natural Abundance	Isotope	Mass	Natural Abundance
^1H	1.0078	99.99%	^{31}P	30.9737	100
^2H	2.0141	0.015	^{32}S	31.9721	95
^{12}C	12	98.89	^{33}S	32.9715	0.76
^{13}C	13.0034	1.11	^{34}S	33.9679	4.22
^{14}N	14.0031	99.64	^{36}S	35.9671	0.02
^{15}N	15.0001	0.36			
^{16}O	15.9949	99.76			
^{17}O	16.9991	0.04			
^{18}O	17.9992	0.2			

70



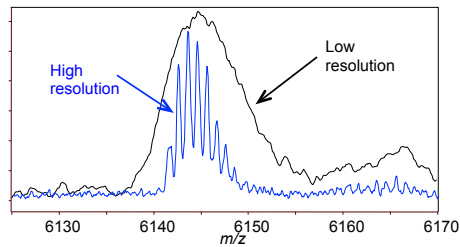
What does the isotopic distributions tell us?



Slide adapted from lecture material developed at the University of Lund

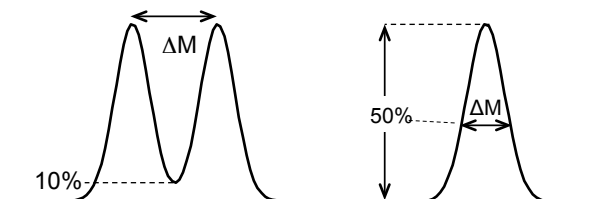
73

Resolution and Resolving Power (RP): terms used interchangeably



The smallest mass difference (ΔM) between peaks such that the valley between them is a specified fraction of the peak height

Full Width Half Maximum (FWHM): Width of a single peak measured at 50% peak apex.



Slide adapted from lecture material developed by A. Burlingame, S. Guan, and M. Baldwin (UCSF) entitled "Mass Spectrometry and Proteomics"

74

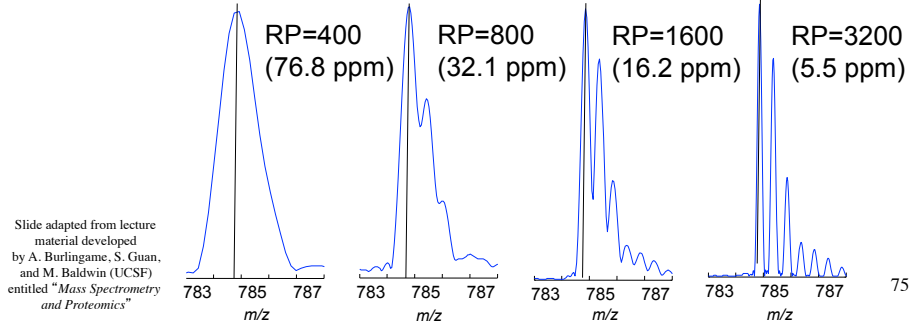
Mass Accuracy (MA) - the difference between the experimental mass (M_{exp}) and the theoretical value (M_{calc}), calculated from elemental composition.

$$MA = \frac{M_{exp} - M_{calc}}{M_{calc}} \quad (\text{ppm for high resolution MS})$$

$$M_{exp} = 1569.684, \quad M_{calc} = 1569.6696 \quad \text{accuracy} = 9.2 \text{ ppm}$$

High resolution means better mass accuracy

m/z :	784.775	784.860	784.848	784.830
M:	1569.549	1569.720	1569.695	1569.661



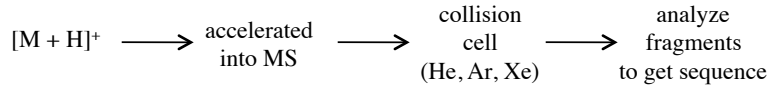
Peptide Mass Fingerprinting : Proteins (or peptides) are digested in a predictable way and the masses of the resulting peptide fragments are unique enough to identify the protein.

Requires a database of known sequences and search software to compare (score) the experimentally observed masses with the calculated masses in the database .

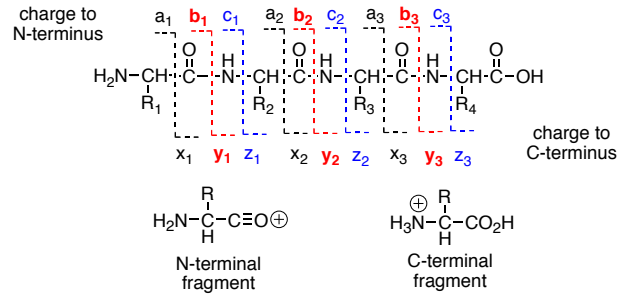
$m/z = 1529 \pm 1 \text{ Da}$	478	peptide fragments from mouse/human genome
1529.7 ± 0.1	164	
1529.73 ± 0.01	25	
1529.7340 ± 0.001	4	
1529.7348 ± 0.0001	2	

Many peptides and proteins give multiply charged ions

CID: collision induced dissociation

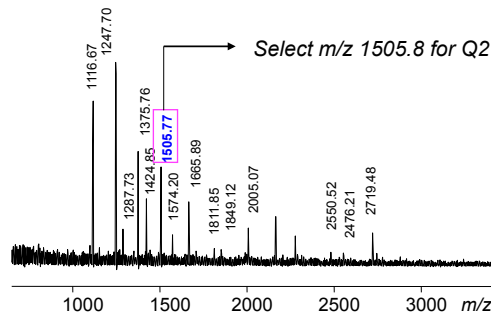
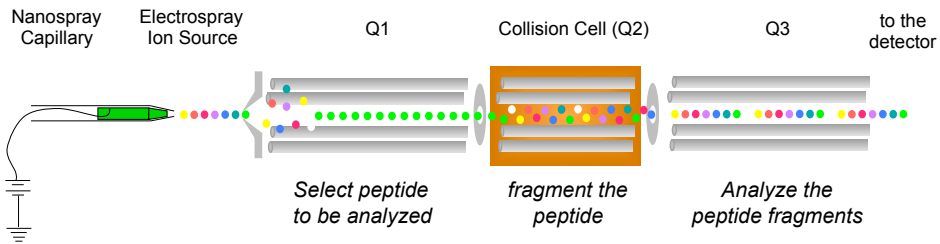


Collision of the $[M+H]^+$ ion with the gas causes it to fragment, analysis of these fragments ions gives sequence information

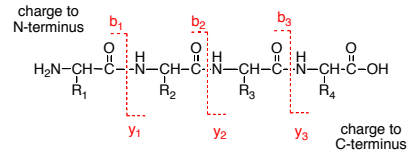


77

Peptide sequencing by tandem mass spectrometry

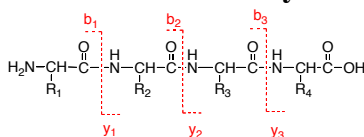


Peptides fragment in a predictable manner



78

Amino Acids Sorted by Mass



		average	exact	<u>-HN-CHR-CO</u>
Glycine	G	75.07	75.03	57.1
Alanine	A	89.10	89.05	71.1
Serine	S	105.09	105.04	87.1
Proline	P	115.13	115.05	97.1
Valine	V	117.15	117.08	99.1
Threonine	T	119.12	119.06	101.1
Cysteine	C	121.16	121.02	103.1
Isoleucine	I	131.18	131.09	113.2
Leucine	L	131.18	131.09	113.2
Asparagine	N	132.12	132.05	114.1
Aspartic Acid	D	133.11	133.04	115.1
Glutamine	Q	146.15	146.07	128.2
Lysine	K	146.19	146.11	128.1
Glutamic Acid	E	147.13	147.13	129.1
Methionine	M	149.21	149.05	131.2
Histidine	H	155.16	155.02	137.1
Phenylalanine	F	165.19	165.19	147.2
Arginine	R	174.20	174.11	156.2
Tyrosine	Y	181.19	181.07	163.2
Tryptophan	W	204.23	204.09	186.2

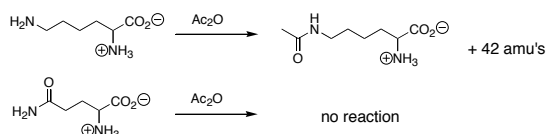
79

Some ambiguities with MS sequencing

leucine (L) vs isoleucine (I): difficult to distinguish, must look at fragmentation of the sidechain



lysine (K, m/z=128.09) vs glutamine (Q, m/z = 128.06)



gly (G) + gly (G) = 114.04 = asn (N) = 114.04

ala (A) + gly (G) = 128.06 = gln (Q) = 128.06 = lys (K) = 128.09

gly (G) + val (V) = 156.09 = arg (R) = 156.10

ala (A) + asp (D) = glu (E) + gly (G) = 186.06 = trp (W) = 186.08

ser (S) + val (V) = 186.1 = trp (W) = 186.08

80

