

# Periodic Table of the Elements

1 I A												18 VIII A	
1s	<b>1</b> <b>H</b> hydrogen 1.008	<b>2</b> <b>II A</b>	atomic # → <b>29</b> +2,1 ← ions commonly formed atomic symbol → <b>Cu</b> English element name → <b>copper</b> ← atomic mass (rounded)										<b>2</b> <b>He</b> helium 4.003
2s	<b>3</b> <b>Li</b> lithium 6.941	<b>4</b> <b>Be</b> beryllium 9.012	3 <b>III B</b> 4 <b>IV B</b> 5 <b>V B</b> 6 <b>VI B</b> 7 <b>VII B</b> 8 <b>VIII B</b> 9 <b>VIII B</b> 10 <b>VIII B</b> 11 <b>I B</b> 12 <b>II B</b>										<b>5</b> <b>B</b> boron 10.81 <b>6</b> <b>C</b> carbon 12.01 <b>7</b> <b>N</b> nitrogen 14.01 <b>8</b> <b>O</b> oxygen 16.00 <b>9</b> <b>F</b> fluorine 19.00 <b>10</b> <b>Ne</b> neon 20.18
3s	<b>11</b> <b>Na</b> sodium 22.99	<b>12</b> <b>Mg</b> magnesium 24.31	Gases    Liquids    Metalloids										<b>13</b> <b>Al</b> aluminum 26.98 <b>14</b> <b>Si</b> silicon 28.09 <b>15</b> <b>P</b> phosphorus 30.97 <b>16</b> <b>S</b> sulfur 32.07 <b>17</b> <b>Cl</b> chlorine 35.45 <b>18</b> <b>Ar</b> argon 39.95
4s	<b>19</b> <b>K</b> potassium 39.10	<b>20</b> <b>Ca</b> calcium 40.08	<b>21</b> <b>Sc</b> scandium 44.96 <b>22</b> <b>Ti</b> titanium 47.87 <b>23</b> <b>V</b> vanadium 50.94 <b>24</b> <b>Cr</b> chromium 52.00 <b>25</b> <b>Mn</b> manganese 54.94 <b>26</b> <b>Fe</b> iron 55.85 <b>27</b> <b>Co</b> cobalt 58.93 <b>28</b> <b>Ni</b> nickel 58.69 <b>29</b> <b>Cu</b> copper 63.55 <b>30</b> <b>Zn</b> zinc 65.41	<b>31</b> <b>Ga</b> gallium 69.72 <b>32</b> <b>Ge</b> germanium 72.64 <b>33</b> <b>As</b> arsenic 74.92 <b>34</b> <b>Se</b> selenium 78.96 <b>35</b> <b>Br</b> bromine 79.90 <b>36</b> <b>Kr</b> krypton 83.80									
5s	<b>37</b> <b>Rb</b> rubidium 85.47	<b>38</b> <b>Sr</b> strontium 87.62	<b>39</b> <b>Y</b> yttrium 88.91 <b>40</b> <b>Zr</b> zirconium 91.22 <b>41</b> <b>Nb</b> niobium 92.91 <b>42</b> <b>Mo</b> molybdenum 95.94 <b>43</b> <b>Tc</b> technetium 98 <b>44</b> <b>Ru</b> ruthenium 101.1 <b>45</b> <b>Rh</b> rhodium 102.9 <b>46</b> <b>Pd</b> palladium 106.4 <b>47</b> <b>Ag</b> silver 107.9 <b>48</b> <b>Cd</b> cadmium 112.4	<b>49</b> <b>In</b> indium 114.8 <b>50</b> <b>Sn</b> tin 118.7 <b>51</b> <b>Sb</b> antimony 121.8 <b>52</b> <b>Te</b> tellurium 127.6 <b>53</b> <b>I</b> iodine 126.9 <b>54</b> <b>Xe</b> xenon 131.3									
6s	<b>55</b> <b>Cs</b> cesium 132.9	<b>56</b> <b>Ba</b> barium 137.3	<b>71</b> <b>Lu</b> lutetium 175.0 <b>72</b> <b>Hf</b> hafnium 178.5 <b>73</b> <b>Ta</b> tantalum 180.9 <b>74</b> <b>W</b> tungsten 183.8 <b>75</b> <b>Re</b> rhenium 186.2 <b>76</b> <b>Os</b> osmium 190.2 <b>77</b> <b>Ir</b> iridium 192.2 <b>78</b> <b>Pt</b> platinum 195.1 <b>79</b> <b>Au</b> gold 197.0 <b>80</b> <b>Hg</b> mercury 200.6	<b>81</b> <b>Tl</b> thallium 204.4 <b>82</b> <b>Pb</b> lead 207.2 <b>83</b> <b>Bi</b> bismuth 209.0 <b>84</b> <b>Po</b> polonium 209 <b>85</b> <b>At</b> astatine 210 <b>86</b> <b>Rn</b> radon 222									
7s	<b>87</b> <b>Fr</b> francium 223	<b>88</b> <b>Ra</b> radium 226	<b>103</b> <b>Lr</b> lawrencium 262 <b>104</b> <b>Rf</b> rutherfordium 261 <b>105</b> <b>Db</b> dubnium 262 <b>106</b> <b>Sg</b> seaborgium 266 <b>107</b> <b>Bh</b> bohrium 264 <b>108</b> <b>Hs</b> hassium 277 <b>109</b> <b>Mt</b> meitnerium 268 <b>110</b> <b>Ds</b> darmstadtium 281 <b>111</b> <b>Rg</b> roentgenium 272 <b>112</b> <b>Cn</b> copernicium 285	<b>113</b> <b>Nh</b> nihonium 284 <b>114</b> <b>Fl</b> flerovium 289 <b>115</b> <b>Mc</b> moscovium 288 <b>116</b> <b>Lv</b> livermorium 292 <b>117</b> <b>Ts</b> tennessine 293 <b>118</b> <b>Og</b> oganeson 294									
lanthanides (rare earth metals)		<b>57</b> <b>La</b> lanthanum 138.9 <b>58</b> <b>Ce</b> cerium 140.1 <b>59</b> <b>Pr</b> praseodymium 140.9 <b>60</b> <b>Nd</b> neodymium 144.2 <b>61</b> <b>Pm</b> promethium 145 <b>62</b> <b>Sm</b> samarium 150.4 <b>63</b> <b>Eu</b> europium 152.0 <b>64</b> <b>Gd</b> gadolinium 157.3 <b>65</b> <b>Tb</b> terbium 158.9 <b>66</b> <b>Dy</b> dysprosium 162.5 <b>67</b> <b>Ho</b> holmium 164.9 <b>68</b> <b>Er</b> erbium 167.3 <b>69</b> <b>Tm</b> thulium 168.9 <b>70</b> <b>Yb</b> ytterbium 173.0											
actinides		<b>89</b> <b>Ac</b> actinium 227 <b>90</b> <b>Th</b> thorium 232.0 <b>91</b> <b>Pa</b> protactinium 231.0 <b>92</b> <b>U</b> uranium 238.0 <b>93</b> <b>Np</b> neptunium 237 <b>94</b> <b>Pu</b> plutonium 239 <b>95</b> <b>Am</b> americium 243 <b>96</b> <b>Cm</b> curium 247 <b>97</b> <b>Bk</b> berkelium 247 <b>98</b> <b>Cf</b> californium 251 <b>99</b> <b>Es</b> einsteinium 252 <b>100</b> <b>Fm</b> fermium 257 <b>101</b> <b>Md</b> mendelevium 258 <b>102</b> <b>No</b> nobelium 259											